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pOH, [H⁺],

[H₃O⁺], [OH⁻] of

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and Buffer

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pK_b H⁺ OH⁻

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EQUILIBRIA

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in Water Water

(H₂O) - the

most important

molecule on

earth. Even in

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pure water,
there are small
amounts of ions
from the

equilibrium

below ("self-

ionization of

water" or "auto-

ionization.

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model of acids

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and bases states that an acid contains the element hydrogen and forms ions of this element when it is dissolved in water. a base contains the hydroxide group and dissociates to produce hydroxide ions

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in aqueous
solution

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PLAY. A neutral solution contains more hydroxide ions than hydrogen ions. False. An aqueous solution of acid causes red litmus paper to turn blue.

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typical
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reaction, if the
reacting acid is
strong, the base
produced is
strong.

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replacement rxn
Rxn in which an
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in an aqueous
solution react
to produce a

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Section 118.18.1
Figure 18.1

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flourish in rich, moist soil that is

moderately acidic, whereas *semper-vivum*, commonly called hen and chicks, grows best in drier, slightly basic soil.

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ending of the

alkane to -oic

acid. i.e. a 4

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from butane to
butanoic acid
Acid salts are
named by
changing the
acid ending to
-oate i.e the 4
carbon acid goes
from butanoic
acid to sodium
butanoate Esters
are named by
changing the
ending of the

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alkane to -oate

and by using

alkyl name i.e.

the 4 carbon

ester below is

methyl

butanoate(methyl

for OCH_3

portion) $\text{O} \quad \text{O}$

Anhydrides are

dehydrated

esters that

replace -oic

acid with word

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anhydride Acyl

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most important

influences on
Enlightenment
thought were
who? John Locke
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2. After 1688,
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permitted
religious
toleration to

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which groups?

Lutherans, Jews,
and Muslims 3.

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Reactions. acid,

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H_3PO_4 , are triprotic acids. Most of the phosphoric acid produced by the chemical industry is used to make fertilizers and detergents, but it is also used to make pharmaceuticals,

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