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a.) C_5H_{12} . b.) $C_{10}H_{22}$. c.) C_2H_2 .

2. 45. Ethanol (CH_3CH_2OH) and propane (C_3H_8) have similar molar masses. Does that mean they have the same boiling point? Explain. 46. For each of the pairs below, circle the substance with the higher boiling point: a.) I_2 and F_2 . b.) HF and LiF .

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c.) CH_2O_2 and CCl_4 . 47.

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C 12.1 Pure substances and mixtures

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181. 1.Answer. In advertisement, a

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Answers product added nothing to it is called as pure. For example: a pure orange juice without additives. While in chemistry- A pure substance is one that is made up of just one substance i.e either element or ...

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Chapter 12 Part B1: Additional Alkyl and Aryl Radicals 2 points Draw the structural or semi-structural formulas for all the alkyl/aryl radicals of the following. Place a X in place of the hydrogen which is removed to make the radical: (Not in Chapter 12-use the

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Chapter 12 – Introduction to Organic
Chemistry: Hydrocarbons Section 12.1
– Alkanes Goal: Identify properties
characteristic of organic or inorganic

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Answers. Summary: Organic compounds: always contain carbon and hydrogen (though sometimes other nonmetals as well.) Inorganic compounds: all other compounds. Often ionic (metal + nonmetal)

Chapter 12 Introduction to Organic

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Answers: Hydrocarbons

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(12) disulfide bridges; salt bridges; H-
bonding; dispersion forces; (metal-ion
coordination is the 5th) (13) Cysteine

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(Cys) is the amino acid that can form disulfide bridges. (14) The acidic amino acids, aspartic acid (Asp) & glutamic acid (Glu) and the basic amino acids, lysine (Lys), arginine (Arg), and histidine (His) can form salt bridges because their side chains are ionized.

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