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*Answers - Periodic Table, Elements,
Metric System \u0026amp; Unit Conversion*

Chapter 11 (Properties of Solutions)

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Periodic Table Explained: Introduction

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Intermolecular Forces - Hydrogen Bonding, Dipole-Dipole, Ion-Dipole, London Dispersion Interactions

Stoichiometry - Limiting & Excess Reactant, Theoretical & Percent Yield - Chemistry

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~~Chemistry, ch 11, Classification of~~
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Exercise Short Question Answer -

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compounds. number of carbon atoms for

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methane. number of carbon atoms for
ethane. number of carbon atoms for
propane.-The formulas of organic
compounds are written with carbon fir....

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Chapter 11 187 Exercise 11.3 – Equation

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Stoichiometry: Iron is combined with carbon in a series of reactions to form pig iron, which is about 4.3% carbon.

$$2C + O_2 \rightarrow 2CO$$
$$Fe_2O_3 + 3CO \rightarrow 2Fe + 3CO_2$$

2CO + C (in iron) \rightarrow CO₂

Pig iron is easier to shape than pure iron, and the presence of carbon lowers its melting point

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Answers to the chapter, and check your answers with the solutions provided in this chapter of the study guide. Ask for help if you need it.

~~Chapter 11 Modern Atomic Theory - An Introduction to Chemistry~~

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Chapter 1 Basic Concepts of Chemistry
and Chemical Calculations; Chapter 2
Quantum Mechanical Model of Atom;
Chapter 3 Periodic Classification of
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Thermodynamics

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AP Chemistry – Solutions, Chapter 11/4b

FAQ. Q: What are hydrophobic and hydrophilic? A: Hydrophobic is a “fear of water”. Nonpolar substances are typically

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Hydrophobic. They are not attracted to water. Hydrophilic on the other hand are “water lovers”. These compounds are attracted to water, and therefore are commonly ionic or polar.

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