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Five Types of Chemical Reaction

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Worksheet Answers. Balance the following reactions and indicate which of the six types of chemical reaction are being represented:

- a) $2. \text{NaBr} + 1. \text{Ca(OH)}_2 \rightarrow 1. \text{CaBr}_2 + 2. \text{NaOH}$ Type of reaction: double displacement.
- b) $2. \text{NH}_3 + 1. \text{H}_2\text{SO}_4 \rightarrow 1. (\text{NH}_4)_2\text{SO}_4$

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Type of reaction: synthesis. c) 4.
 $C_5H_9O + 27. O_2 \rightarrow 20. CO_2 + 18.$
 H_2O Type of reaction:

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to show how substances in a chemical reaction interact. to shorten the explanation of a chemical reaction. to help keep track of atoms in a chemical reaction. Nitrogen gas reacts with hydrogen to produce ammonia.

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Use the chemical equation to complete the statements. $\text{N}_2 + 3\text{H}_2 \rightarrow 2\text{NH}_3$.

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Reactions Worksheets Answers

We count chemical entities in a reaction by measuring macroscopic observables like mass, volume, temperature and pressure. What we measure depends on the state of matter, but we need to relate a

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measurable property to the number of particles. In this chapter we will learn how to do this for solid substances and for solutes in a solution.

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